

Rapid and facile synthesis of spiro[indole-3,3'-[1,2,4]triazol]-2(1H)-ones

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Various 4'-aryl-5'-phenyl-2',4'-dihydrospiro[indole-3,3'-[1,2,4]triazol]-2(1H)-ones have been synthesised by reaction of amidrazones with 2-(2-oxoindolin-3-ylidene)malononitrile.

Keywords: amidrazones, 2-(2-oxoindolin-3-ylidene)malononitrile and 4'-aryl-5'-phenyl-2',4'-dihydrospiro[indole-3,3'-[1,2,4]triazol]-2(1H)-ones

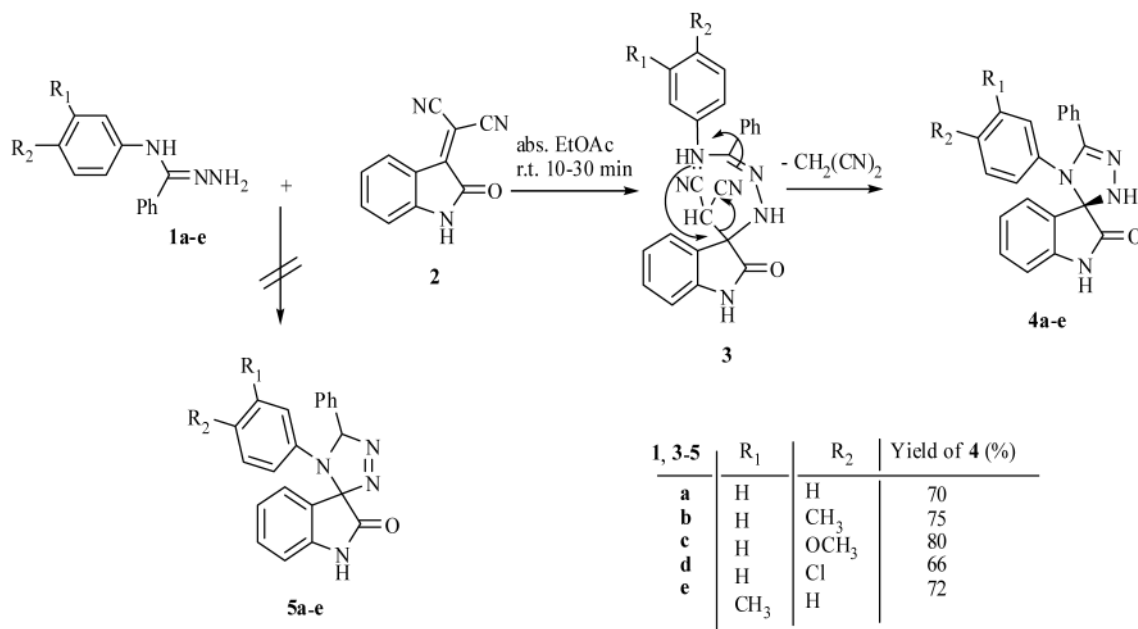
1,2,4-Triazole moieties have been incorporated into a wide variety of therapeutically interesting drug candidates including anti-inflammatory, CNS stimulants, sedatives, anti-anxiety compounds, antimicrobial agents,¹⁻³ and antimycotics such as voriconazole, fluconazole and intraconazole.^{4,5} There are marketed drugs containing the 1,2,4-triazole group which are known to have antimycotic effects *e.g.*, alprazolam⁶ and triazolam.¹⁷ Amidrazones are an important class of amidines, since they can be used as synthons for 1,2,4-triazolo compounds and other heterocyclic systems.^{7,8} Amidrazones possessing three nitrogen nucleophiles are essential components of molecules exhibiting high biological activities, *e.g.*, lamotrigine and sildenafil (Viagra).⁹ Previously investigations described the reactions of amidines and their analogues with π -acceptors and various heterocycles.¹⁰⁻¹² Aly *et al* have recently reviewed the use of hydrazinecarbothioamide, thiocarbonylhydrazide and other derivatives in the synthesis of heterocycles.¹³ Interestingly, a very convenient procedure to synthesise 1,2,4-triazoles in one step has been demonstrated, by a rapid reaction of amidrazones with 2-(1,3-dioxindan-2-ylidene)malononitrile.¹⁴ Accordingly, we have now investigated the reactions of amidrazones **1a-e** with 2-(2-oxoindolin-3-ylidene)malononitrile (**2**) as electron π -deficient acceptor, aiming to obtain heterocyclic compounds which might have biological activities.

Results and discussion

Scheme 1 outlines the reaction of **1a-e** with **2** in dry ethyl acetate under N₂ atmosphere. The reaction proceeded in a few

minutes to yield, after chromatographic purification and recrystallization, compounds **4a-e** (66–80%). We chose amidrazones **1a-e** having aryl groups with electron donating and withdrawing substituents on the benzene ring, in order to examine the effect of electron demand on the course of reaction.

The structure assignment of compounds **4a-e** is well established using traditional spectroscopic tools such as IR, NMR (¹H, ¹³C) and mass spectra, in addition to elemental analyses. Compounds **4a-e** showed IR absorption peaks for NH groups at $\nu_{\max} = 3350\text{--}3330\text{ cm}^{-1}$ along with broad bands at $\nu_{\max} = 1680\text{--}1695\text{ cm}^{-1}$ corresponding to the carbonyl groups. The mass spectra of compounds **4a-e** showed the molecular peaks, albeit not as the base peaks. For example, **4a** showed a molecular ion peak at $m/z = 340$ (68%), and the elemental analysis confirmed the molecular formula as C₂₁H₁₆N₄O. The mass spectra of all compounds **4a-e** contain similar fragments at $m/z = 91, 113, \text{ and } 160$ (see Fig. 1). The ¹H NMR spectrum of **4a** showed the fused phenyl protons at $\delta_{\text{H}} = 7.90$ (t, 2 H, $J = 7.0$ Hz) and 7.85 (d, 2 H, $J = 7.8$ Hz), whilst two NH-protons were observed as broad singlets at $\delta_{\text{H}} = 6.5$ and 8.0. The observed ¹³C NMR signals of **4a** confirmed the proposed structure of **4a** by the appearance of *spiro*-C at $\delta_{\text{C}} = 89.0$, whilst the C-2 and C=N carbons resonated at $\delta_{\text{C}} = 168.2$ and 156.6 respectively. The chemical shifts (δ_{C}) of some distinctive carbons of compounds **4a** and **4b** are shown in Fig. 2. The observation of two NH-signals, but no CH-proton or carbon signals, is consistent with structures **4a-e** but excludes the isomeric triazoles **5a-e** (Scheme 1).



Scheme 1 Synthesis of new 4'-aryl-5'-phenyl-2',4'-dihydrospiro[indole-3,3'-[1,2,4]triazol]-2(1H)-ones **4a-e**.

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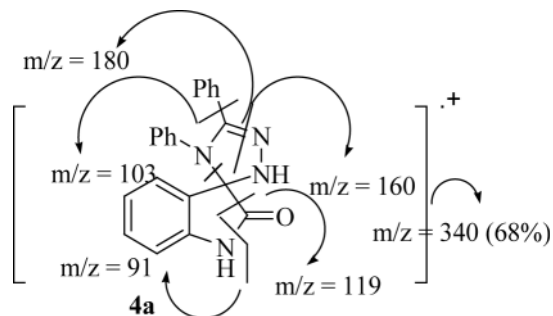


Fig. 1 Mass fragmentation patterns of compound **4a**.

The molecular modelling (MM2)¹⁵ of **4a**, as an example, indicated that the bond distance between the NH proton and the oxygen of the carbonyl group (C-2) is ~ 1.7 Å enabling them to form an intramolecular hydrogen bond (Fig. 3). The calculated (CHEM-NMR) chemical shift of the hydrazine-NH proton is $\delta_{\text{H}} = 4.0$, whereas its experimental value is $\delta_{\text{H}} = 6.50$, in accord with deshielding due to the hydrogen bond just postulated. The angle between the plane N2–C3–N4 of the triazole ring and the plane C2–C3–C3a of the indole is calculated to be 105° .

Mechanistically, the reaction of **2** with **1a–e** is considered to involve addition of the hydrazino-NH₂ of **1** to the ylidene carbon in **2** to produce the intermediate **3** (Scheme 1). Subsequently, elimination of malononitrile, followed by another nucleophilic attack of the imino-NH *via* amidine-like addition to the electrophilic carbon, gives the stable heterocyclic compounds **4a–e** (Scheme 1). The presence of electron donating substituents such as methyl and methoxy *para* to the proximal nitrogen atom in **1a–e** (R₂) increased the yields of products **4a–e**, consistent with nucleophilic attack by this nitrogen.

Experimental

Melting points are uncorrected values. ¹H NMR and ¹³C NMR spectra (Bruker AM 400, ¹H: 400.13 MHz, ¹³C: 100.6 MHz) were obtained from CDCl₃ solutions; the chemical shifts (δ) are given relative to internal standard TMS, and coupling constants (*J*) are in Hz. For preparative thin layer chromatography (PLC), glass plates (20 × 48 cm) were covered with a slurry of silica gel Merck PF₂₅₄ and air dried using the solvents listed for development. Zones are detected by quenching of indicator fluorescence upon exposure to 254 nm UV light. Elemental analyses were carried out in the Microanalysis Center of the Institut für Anorganische Chemie, Technische Universität Braunschweig. Mass spectroscopy was performed with a Finnigan Mat 8430 spectrometer at 70 eV, Institute of Organic Chemistry, TU-Braunschweig, Germany. IR spectra were obtained on Shimadzu 470 spectrophotometer using potassium bromide pellets; absorption maxima (ν_{max}) are reported in cm⁻¹.

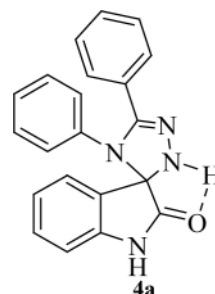


Fig. 3 Hydrogen bond in compound **4a**.

Amidrazones **1a–e** were prepared according to ref. 7. 2-(2-Oxoindolin-3-ylidene)malononitrile (**2**) was prepared¹⁶ from isatin (Merck).

Reaction of **1a–e** with 2-(2-oxoindolin-3-ylidene)malononitrile (**2**)

General procedure

A 250 mL two-necked bottom flask was flame-dried under N₂ atmosphere and then cooled to room temperature. Absolute ethyl acetate (100 mL) containing a mixture of **1a–e** (1 mmol) and **2** (0.195 g 1 mmol) was added to this flask. The reaction mixture was further stirred at room temperature for 10–30 min until the starting materials were consumed (the reaction progress was monitored by TLC analysis). The solvent was removed under vacuum and the residue was separated by PLC (silica gel, toluene: ethyl acetate 5: 1). The obtained products **4a–e** were recrystallised from the stated solvents.

4',5'-Diphenyl-2',4'-dihydrospiro[indole-3,3'-[1,2,4]triazol]-2(1H)-one (4a): Colourless crystals (0.24 g, 70%), m.p. (methanol) 160–162 °C. IR: $\nu_{\text{max}} = 3350$ (NH), 3090–3010 (ArCH), 1695 (C=O), 1580 (C=C). ¹H NMR: $\delta_{\text{H}} = 8.0$ (br, s, 1 H, NH, exchangeable with D₂O), 7.90 (t, 2 H, *J* = 7.0), 7.85 (d, 2 H, *J* = 7.8), 7.56–7.20 (m, 10 H, ArH), 6.50 (br, s, 1 H, NH, exchangeable with D₂O). ¹³C NMR: $\delta_{\text{C}} = 168.2$ (C=O), 156.6 (C=N), 142.0, 132.0, 131.0, 130.6 (ArC), 128.7, 128.4 (ArCH), 128.2, 127.4 (ArCH-*p*), 127.0, 126.8 (Ar2CH-*o*), 126.2, 126.0 (Ar2CH-*m*), 125.8, 118.6 (ArCH), 89.0 (spiro-C). MS (70 eV, EI): *m/z* = 340 (68), 180 (28), 160 (24), 113 (26), 103 (40), 194 (60), 91 (100), 77 (28). Anal. Calcd for C₂₁H₁₆N₄O (340.381): C, 74.10; H, 4.74; N, 16.46. Found: C, 73.89; H, 4.70; N, 16.38.

4'-(4''-Methylphenyl)-5'-phenyl-2',4'-dihydrospiro[indole-3,3'-[1,2,4]triazol]-2(1H)-one (4b): Colourless crystals (0.27 g, 75%), m.p. (ethanol) 228–230 °C. IR: $\nu_{\text{max}} = 3330$ (NH), 3090–3015 (ArCH), 2980–2870 (aliph.-CH), 1680 (C=O), 1610 (C=N), 1580 (C=C). ¹H NMR: $\delta_{\text{H}} = 8.20$ (br, s, 1 H, NH, exchangeable with D₂O), 8.10 (t, 2 H, *J* = 7.0), 7.92 (d, 2 H, *J* = 7.8), 7.66–7.10 (m, 9 H, ArH), 6.30 (br, s, 1 H, NH, exchangeable with D₂O), 2.32 (s, 3 H, CH₃). ¹³C NMR: $\delta_{\text{C}} = 168.8$ (C=O), 156.6 (C=N), 142.2, 134.2, 132.4, 131.4, 131.2 (ArC), 129.0, 128.6 (ArCH), 128.4 (ArCH-*p*), 128.0, 127.6 (Ar2CH-*o*), 127.2, 127.0 (Ar2CH-*m*), 126.5, 119.0 (ArCH), 89.6 (spiro-C), 22.1 (CH₃-Ar). MS (70 eV, EI): *m/z* (%) = 354 (60), 326 (22), 263 (16), 251 (20), 160 (20), 113 (12), 110 (100), 91 (60), 91 (34), 77 (30).

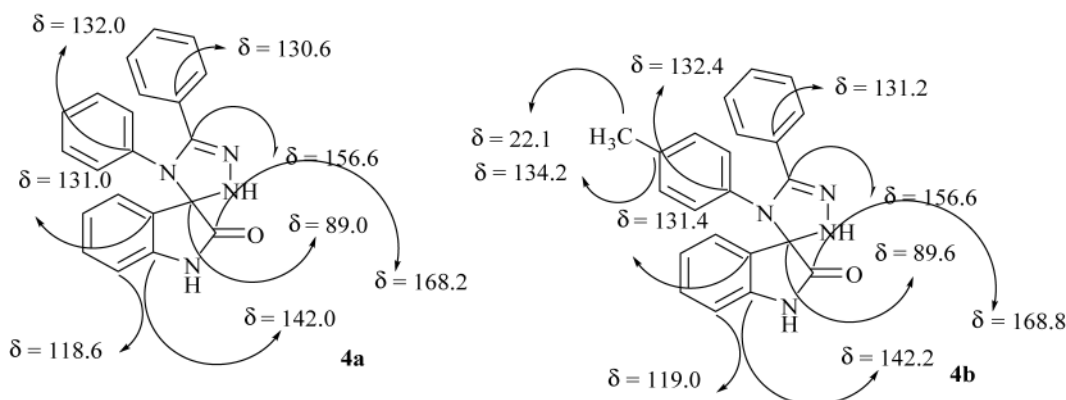


Fig. 2 Assignment carbon signals and their δ values of compounds **4a** and **4b**.

Anal. Calcd for C₂₂H₁₈N₄O (354.407): C, 74.56; H, 5.12; N, 15.81. Found C, 74.40; H, 5.10; N, 15.70.

4'-(4''-Methoxyphenyl)-5'-phenyl-2',4'-dihydrospiro[indole-3,3'-[1,2,4]triazol]-2(1H)-one (**4c**): Colourless crystals (0.3 g, 80%), m.p. (methanol) 190 °C. IR: ν_{\max} = 3338 (NH), 3087–3010 (ArCH), 2987–2860 (aliph.-CH), 1680 (C=O), 1610 (C=N), 1586 (C=C). ¹H NMR: δ_{H} = 8.30 (br, s, 1 H, NH, exchangeable with D₂O), 8.10 (d, 2 H, *J* = 7.8), 7.80 (d, 2 H, *J* = 7.8), 7.70–7.12 (m, 7 H, ArH), 6.90 (m, 2 H, ArH), 6.40 (br, s, 1 H, NH, exchangeable with D₂O), 3.90 (s, 3 H, OCH₃). ¹³C NMR: δ_{C} = 169.2 (C=O), 158.0 (C=N), 156.0, 134.4, 132.0, 131.2, 131.0 (ArC), 129.2, 128.8 (ArCH), 128.2 (ArCH-*p*), 127.4 (Ar2CH-*o*), 127.0, 126.8 (Ar2CH-*m*), 122.0 (Ar2CH-*o*), 126.5, 119.0 (ArCH), 90.0 (spiro-C), 51.0 (CH₃-Ar). MS (70 eV, EI): *m/z* = 370 (56), 342 (22), 327 (22), 249 (30), 210 (77), 160 (24), 113 (28), 110 (100), 91 (22), 77 (20). Anal. Calcd for C₂₂H₁₈N₄O₂ (370.407): C, 71.34; H, 4.90; N, 15.13. Found: C, 71.30; H, 4.80; N, 15.08.

4'-(4''-Chlorophenyl)-5'-phenyl-2',4'-dihydrospiro[indole-3,3'-[1,2,4]triazol]-2(1H)-one (**4d**): Pale yellow crystals (0.3 g, 66%), m.p. (ethyl acetate) 220 °C. IR: ν_{\max} = 3334 (NH), 3098–3040 (ArCH), 1682 (C=O), 1612 (C=N), 1580 (C=C). ¹H NMR: δ_{H} = 8.40 (br, s, 1 H, NH, exchangeable with D₂O), 7.40–6.80 (m, 13 H, ArH), 6.10 ((br, s, 1 H, NH, exchangeable with D₂O). ¹³C NMR: δ_{C} = 168.8 (C=O), 156.2 (C=N), 132.0, 131.8, 130.2, 130.0 (ArC), 128.5, 128.2 (ArCH), 128.0 (ArCH-*p*), 126.8, 126.6 (Ar2CH-*o*), 126.2 (Ar2CH-*m*), 125.4 (Ar2CH-*l*), 122.0 (ArC), 125.6, 118.6 (ArCH), 89.2 (spiro-C). MS (70 eV, EI): *m/z* = 376 [M+2] (38), 374 (100), 330 (22), 271 (16), 214 (30), 160 (32), 91 (80), 114 (28), 113 (32), 77 (24). Anal. Calcd for C₂₁H₁₅N₄ClO (374.83): C, 67.29; H, 4.04; N, 14.95. Found: C, 67.12; H, 4.00; N, 15.00.

4'-(3''-Methylphenyl)-5'-phenyl-2',4'-dihydrospiro[indole-3,3'-[1,2,4]triazol]-2(1H)-one (**4e**): Colourless crystals (0.26 g, 72%), m.p. (ethanol) 200 °C. IR: ν_{\max} = 3334 (NH), 3092–3020 (ArCH), 2960–2860 (aliph.-CH), 1684 (C=O), 1612 (C=N), 1586 (C=C). ¹H NMR: δ_{H} = 8.10 (br, s, 1 H, NH, exchangeable with D₂O), 8.00 (t, 2 H, *J* = 7.4), 7.90 (d, 2 H, *J* = 7.6), 7.70–6.96 (m, 9 H, ArH), 6.20 (br, s, 1 H, NH, exchangeable with D₂O), 2.30 (s, 3 H, CH₃). ¹³C NMR: δ_{C} = 169.0 (C=O), 156.5 (C=N), 142.0, 134.0, 132.2, 131.6, 131.4 (ArC), 129.0, 128.8 (ArCH), 128.6 (ArCH-*p*), 128.2, 127.6 (Ar2CH-*o*), 127.4, 127.2 (Ar2CH-*m*), 126.3, 119.2 (ArCH), 89.2 (spiro-C), 22.4 (CH₃-Ar). MS (70 eV, EI): *m/z* (%) = 354 (56), 326 (20), 263 (14), 251 (24), 160 (24), 113 (22), 110 (100), 91 (46), 91 (30), 77 (28).

Anal. Calcd for C₂₂H₁₈N₄O (354.407): C, 74.56; H, 5.12; N, 15.81. Found C, 74.44; H, 5.08; N, 15.68.

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